Exam. Code : 107402

Subject Code: 2214

B.Sc. Bio-Technology 2nd Semester ORGANIC CHEMISTRY-B

Paper—BT-4

Time Allowed—Three Hours [Maximum Marks—40]

SECTION-A

Note: -- ALL questions are compulsory.

- Alkynes react with ammonical solution of AgNO, to give a white precipitate but alkenes do not give this reaction. Explain.
- How will you convert acetylene to 3-octyne?
- What are the limitations of Williamson's ether synthesis?
- How will you convert ethylene oxide in to 4. 1-Hexanol?
- 5. Complete the following reaction with suitable mechanism:

3114(2517)/STB-14044

(Contd.)

- 6. What combination of carbonyl compound and ylide would you use to prepare styrene?
- 7. Amides are weaker bases than amines, why?
- 8. Why acyl chlorides are easily hydrolysed than amides?
 1×8=8

SECTION-B

Note: — Attempt any *five* questions. All questions carry equal marks.

- 9. Alkynes are less reactive than alkenes towards electrophilic addition reactions. Explain.
- 10. Write down the reaction of but-2-yne with alkali metal in liquid ammonia? Give its mechanism.
- 11. Anisole is prepared by the reaction between sodium phenoxide and methyl bromide and not by the reaction between sodium methoxide and bromobenzene.
- 12. Discuss regioselectivity of ring opening of unsymmetrical oxirane under acidic and basic conditions.
- 13. Provide suitable conditions for the following conversion and provide a suitable mechanism for it:

3114(2517)/STB-14044

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(Contd.)

- 14. With mechanism, discuss how will you distinguish between acetaldehyde and benzaldehyde.
- 15. Write down the base-catalyzed mechanism of hydrolysis of esters.
- 16. With mechanism, discuss Reformatsky reaction.

SECTION-C

Note: - Attempt any two questions. All questions carry equal marks.

17. (a) Complete the following reaction and provide a suitable mechanism:

(b) Discuss various factors responsible for the acidity of terminal alkynes.

18. (a) Write a note on crown ethers. 3

(b) Complete the following reactions:

3114(2517)/STB-14044

(Contd.)

3

19. (a) Provide a suitable mechanism for the following conversion:

(b) Complete the following reaction and provide a suitable mechanism:

20. (a) Complete the following reaction and provide a suitable mechanism:

(b) Provide a suitable mechanism for the following conversion:

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